

A CONCISE SYNTHESIS OF 12-(3'-HYDROXY-n-PROPYL)-DEOXOARTEMISININ

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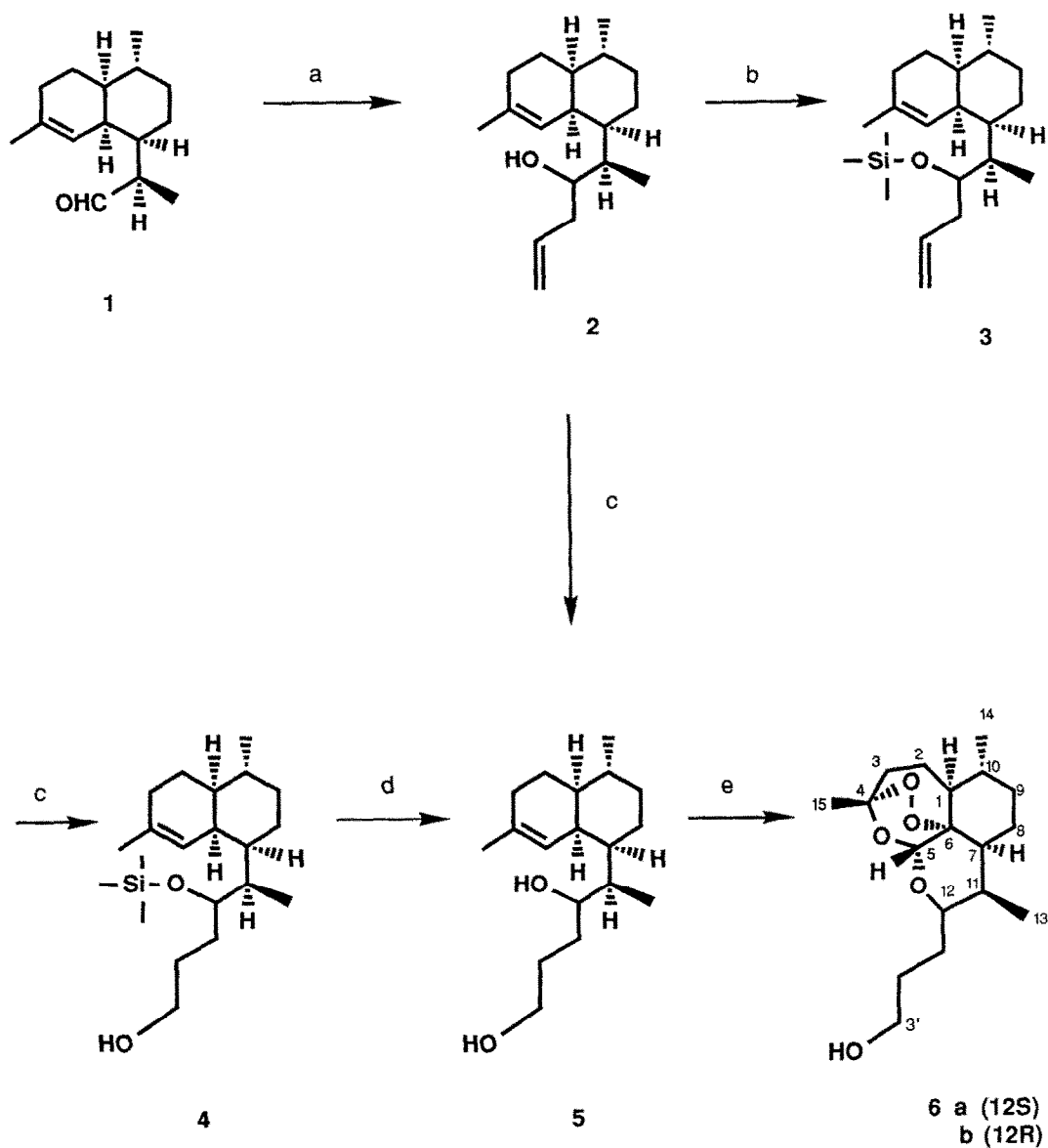
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Abstract: The 12-(3'-hydroxy-n-propyl)-deoxoartemisinin was synthesized from artemisinic acid *via* photooxidative cyclization as a key step. The title compound **6a** is found to show five times greater *in vitro* antimalarial activity compared to artemisinin.

In an ongoing search for new and therapeutically useful antimalarials, we sought to synthesize deoxoartemisinin (deoxoqinghaosu) analog, **6**¹. Artemisinin analogs lacking the carbonyl function at C-12 are projected to possess increased stability, thus longer half-life in the body². However, it is difficult to prepare **6** from artemisinin by direct derivatization at C-12 due to the chemically labile functional groups within the molecule. In this communication, we wish to report a successful synthesis of **6**.

Because of its greater abundance³ and suitable stereochemical configuration, artemisinic (qinghao) acid is a useful chiral synthon for the preparation of artemisinin analogs⁴. A strategy for a conversion of (+)-artemisinic acid into 12-(3'-hydroxy-n-propyl)-deoxoartemisinin, **6** is outlined in Scheme 1. Grignard addition to dihydroartemisinylaldehyde **1**, prepared from artemisinic acid^{1c}, with allylmagnesiumbromide (2.88 eq.) in anhydrous ether (r. t., 30 min.) afforded **2** (12S/12R=1:1) in 62 % yield. Treatment of **2** with t-butyldimethylsilyl chloride (imidazole, DMF, r.t.) gave **3** (93 %) and hydroboration of **3** with 9-BBN in anh. THF (r.t., 24 h.) and subsequent oxidation with 30 % hydrogen peroxide/2N-sodium hydroxide (50 °C, 1 h.) afforded **4** (71 %) with high regioselectivity. The silyl protecting group of the secondary alcohol of **4** was cleanly removed by treatment with tetrabutylammonium fluoride in anh. THF (r.t., 20 h.) to give the diol **5** in high yield (90 %). **5** was also directly prepared from **2** with 9-BBN and subsequently hydrogen peroxide/sodium hydroxide, as described above, in 80 % yield. Chiral photooxidation of **5** with oxygen [irradiation with 450-W medium pressure mercury arc lamp at -78 °C, methylene blue(cat.)] in methylene chloride, followed by acidic Dowex-resin

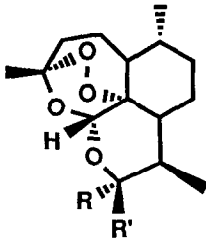
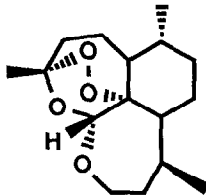
Scheme 1.



- a) allyl bromide, Mg, anh. ether, r.t., 30 min. b) TBDMSCl, imidazole, DMF, r. t., 2 h.
 c) 9-BBN (2.5 eq.), anh. THF, r. t., 24 h. : 30 % H_2O_2 / 2N-NaOH, 50 °C, 1 h.
 d) $n\text{-Bu}_4\text{NF}$, anh. THF, r. t., 20 h. e) oxygen, irradiation at -78 °C, methylene blue (cat.), CH_2Cl_2 , 3 h.: Dowex resin (acidic) (cat.), ether, r. t., 18 h.

catalyzed cyclization (ether, r.t., 18 h.) of the oxygenation intermediates afforded **6a**(12S)⁵ and **6b**(12R)⁵ of natural configuration in 12 % and 11 % yield, respectively⁶. This conversion proves to be a concise synthesis of the title compound **6a** and its 12-epimer, **6b** from artemisinic acid. The assignments of the ¹H NMR and ¹³C NMR signals were made on the basis of 2D-COSY and HETCOR spectra of 12-(3'-hydroxy-n-propyl)-deoxoartemisinin **6a** and **6b**, respectively. The relative configuration at the new chiral centers, C-4, 5, 6, 11, and 12 was unambiguously determined as depicted in **6a** and **6b**, respectively, by utilization of two dimensional nOe (NOSEY) techniques⁷. Table 1 shows *in vitro* assessment of artemisinin-related analogs. **6a** is found to show five times greater *in vitro* antimalarial activity compared to artemisinin against chloroquine-resistant malaria. **6a** is approximately 80 times more active compared to homoartemisinin **9**^{1c}. Due to the lack of the carbonyl function and exo C-O bond at C-12, **6a** is projected to possess increased stability, thus longer half life in the body².

Table 1

		IC ₅₀ (ng/mL) clone of <i>P.falciparum</i>	
	6a : R = (CH ₂) ₃ -OH R' = H	12-(3'-hydroxy-n-propyl)-deoxoartemisinin	0.23
	7 : R, R' = O	artemisinin	1.21
	8 : R = R' = H	deoxoartemisinin	0.15
		9	homodeoxoartemisinin

In conclusion, this communication describes the concise synthesis of 12-(3'-hydroxy-n-propyl)-deoxoartemisinin from artemisinic acid.

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References and Notes:

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5. For **6a**: $[\alpha]_D^{18} = -20.25^\circ$ (c 0.8, CHCl₃): ¹H NMR (CDCl₃, 300 MHz) 5.26(s, 1H, C5-H), 3.69(m, 2H, C3'-H), 3.52 (m, 1H, C12-H), 2.42(m, 1H, C11-H), 2.34(ddd, 1H, J=14.96, 4.02, 1.17 Hz, C3 α -H), 2.03, and 1.98(dq, 1H, J=14.96, 3.0 Hz, C3 β -H), 1.40 (s, 3H, C15-H), 0.95(d, J=6.2 Hz, 3H, C13-H), 0.79(d, J=7.2 Hz, 3H, C14-H): ¹³C NMR (CDCl₃, 75 MHz) 104.18, 92.14, 80.82, 74.05, 63.03, 51.91, 45.97, 37.39, 36.16, 34.16, 30.63, 29.80, 27.64, 25.93, 24.74, 21.44, 20.29, 13.76 ppm. IR (CHCl₃) $\nu_{\max}$ 3440(br.s), 2925, 1505, 1380, 1130, 1100, 1060, 1040, 880 cm^{-1} ; MS(70eV): m/e, 326(M⁺, 53.49), 294(M⁺-O₂, 7.11). For **6b**: $[\alpha]_D^{18} = +47.25^\circ$ (c 0.4, CHCl₃): ¹H NMR (CDCl₃, 300 MHz) 5.33(s, 1H, C5-H), 4.23(m, 1H, C12-H), 3.68 (m, 2H, C3'-H), 2.65(m, 1H, C11-H), 2.32(ddd, 1H, J=14.01, 3.96, 4.02 Hz, C3 α -H), 2.05 and 2.00(dq, 1H, J=14.01, 3.12, C3 β -H), 1.42(s, 3H, C15-H), 0.96(d, J=6 Hz, 3H, C13-H), 0.87 (d, J=7.53 Hz, 3H, C14-H): ¹³C NMR (CDCl₃, 75 MHz) 103.18, 89.20, 81.11, 75.44, 62.77, 52.24, 44.21, 37.48, 36.54, 34.42, 31.22, 30.50, 26.44, 26.00, 24.87, 24.72, 20.16, 12.88: IR (CHCl₃) $\nu_{\max}$ 3440(br.s), 2950, 1450, 1380, 1130, 1070, 1020, 635 cm^{-1} .
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